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Stereoselective Oxindole Synthesis by Palladium-Catalyzed Cyclization Reaction of 2-(Alkynyl)aryl Isocyanates with Amides

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ABSTRACT

A new cyclization reaction occurred on treatment of 2-(alkynyl)aryl isocyanates with amides in the presence of a palladium(0)/diphosphine catalyst to stereoselectively form 3-(amidoalkylidene)oxindoles. A carbon—nitrogen bond as well as a carbon—carbon bond were simultaneously introduced onto the alkyne moiety to construct an oxindole skeleton with stereoselective placement of the amino substituent *cis* to the carbonyl group.

Transition metal-catalyzed C-N bond-forming reactions have been the subject of intense research¹ because of the importance of nitrogen-containing compounds. Oxindoles are often found in bioactive molecules as the key substructure,² which has driven increased interest in exploring new methods for their preparation. In particular, 3-(aminoalkylidene)oxindoles are of significant pharmaceutical value,³ and there-

fore, methods to prepare them in a stereodefined way are in high demand. We report herein a palladium-catalyzed cyclization reaction of 2-(alkynyl)aryl isocyanates⁴ with primary and secondary amides, which produces 3-(amidoalkylidene)oxindoles.⁵ The reaction allows for the intermolecular C—N bond introduction onto the alkyne moiety *cis* to the developing carbonyl group in a stereoselective fashion.

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We previously described the palladium-catalyzed cyclization of 2-(alkynyl)aryl isocyanates with organoboron reagents.6 In this reaction, an oxapalladacyclic intermediate formed by oxidative cyclization undergoes transmetalation with an organoboron species, converting a palladium—oxygen bond into a palladium-carbon bond. It was then envisaged that the use of protic nitrogen nucleophiles in place of organoborons would result in the generation of a palladium—nitrogen bond through ligand substitution, leading to the introduction of a carbon-nitrogen linkage. Thus, 2-(1-hexynyl)phenyl isocyanate (1a, 1.0 equiv) was treated with trifluoroacetamide (2a, 1.1 equiv) in the presence of Pd₂(dba)₃·CHCl₃/dppf (1 mol %; dppf = 1,1'-bis(diphenylphosphino)ferrocene) in toluene (0.05 M) at 100 °C. The reaction reached completion in 3 h, and an extractive workup followed by precipitation from CH₂Cl₂/hexane afforded the 3-(amidoalkylidene)oxindole **3aa** in 83% yield as a single stereoisomer (Z/E = >20: 1,' eq 1).

We propose that the reaction proceeds through the pathway outlined in Scheme 1. Initially, both alkynyl and isocyanato

Scheme 1. Proposed Reaction Pathway

groups of **1a** coordinate to a palladium(0) center prompting oxidative cyclization to form oxapalladacyclic intermediate **B**.⁸ The palladium(II) oxide moiety then acts as a base to

promote ligand substitution by amide 2a and thus, the additional usage of an external base is dispensed with for generation of the palladium(II) amide species \mathbb{C}^9 . Finally, reductive elimination from \mathbb{C} affords the product 3aa, regenerating the palladium(0) catalyst.

As shown in Table 1, primary and secondary alkyl groups were suitable for the substituent at the alkyne terminus

Table 1. Pd(0)-Catalyzed Cyclization Reaction of **1** with Trifluoroacetamide $(2\mathbf{a})^a$

entry	1	R ¹	\mathbb{R}^2	3	yield (%) ^b
1	1b	<i>n</i> -Pr	Н	3ba	86
2	1c	<i>i</i> -Pr	Н	3ca	88
3	1d	cyclopropyl	Н	3da	79
4	1e	t-Bu	Н	3ea	56^c
5	1f	Ph	Н	3fa	99
6	1g	4-MeO-C_6H_4	Н	3ga	93
7	1h	$4-CF_3-C_6H_4$	Н	3ha	86
8	1i	$4-CH_3-C_6H_4$	Н	3ia	93
9	1 j	$2-CH_3-C_6H_4$	Н	3ja	99
10	1 k	3-thienyl	Н	3ka	95
11	11	vinyl	Н	3la	72
12	1m	Ph	Br	3ma	99
13	1n	<i>n</i> -Bu	Cl	3na	74
14	10	n-Bu	OMe	3oa	84
15	1p	n-Bu	CO ₂ Et	3pa	80
16	1q	<i>n</i> -Bu	CN	3qa	82
	CI、 ^	n-Bu	n-Bu	COCF₃ N.	
17		NCO		`H ≻= O	96
		1r	, N	3ra	

 a Conditions: **1** (0.2 mmol), **2a** (0.22 mmol), Pd₂(dba)₃·CHCl₃ (2 μmol, 1 mol %), and dppf (4 μmol, 2 mol %) in toluene (4 mL) at 100 °C for 3 h under Ar. b Isolated yield (stereoisomer ratio = >20:1). c **2a** (0.6 mmol, 3 equiv) was used for 13 h.

(entries 1–3). Even the bulky *tert*-butyl group permitted the reaction, albeit under more forcing conditions, to give the product **3ea** in 56% yield (entry 4). Substrates possessing a wide range of aryl and heteroaryl groups successfully participated in the cyclization reaction (entries 5–10). Vinyl-substituted substrate **1l** was also converted to the product **3la** in good yield (entry 11). Functional groups including halide, ether, ester, and nitrile were tolerated on the aryl group of **1** (entries 12–17). Interestingly, even the bromo group of **1m** remained intact.

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⁽⁷⁾ The stereochemistry of the exocyclic double bond was assigned by a difference NOE study.

⁽⁸⁾ A result supportive for the presumed oxapalladacycle was obtained by a ^{1}H NMR study; when **1a** was treated with a stoichiometric amount of Pd₂dba₃·CHCl₃ and dppf in THF- d_8 at 80 $^{\circ}$ C for 2 h, the signal ($-CH_2C_3H_7$) shifted downfield from 2.45 to 2.76 ppm. However, an attempt to isolate it as solids has been unsuccessful so far.

⁽⁹⁾ Driver, M. S.; Hartwig, J. F. *J. Am. Chem. Soc.* **1996**, *118*, 4206. (10) The substrates with R1 = H and SiMe₃ gave a complex mixture.

A straightforward synthesis of **3aa** on a gram scale was also carried out to demonstrate the practicality of the present method (Scheme 2). Product **3aa** (2.65 g, 8.5 mmol) was

Scheme 2. Synthesis of 3aa on Gram Scale

obtained starting from 2-(1-hexynyl)aniline (5, 1.73 g, 10 mmol) via the corresponding isocyanate 1a without intervention of any chromatographic purification (85% yield over two steps).

Next, hydrolysis of the trifluoroacetylamide group was examined since 3-(free aminoalkylidene)oxindoles **4** have been identified as potent kinase inhibitors. ¹¹ In fact, this group could be easily removed by treatment with the mild base K_2CO_3 , ¹² as exemplified in Table 2.

Table 2. Deprotection Reaction of Trifluoroacetyl Group^a

entry	3	\mathbb{R}^1	\mathbb{R}^2	4	yield ^b (%)
1	3aa	n-Bu	Н	4a	89
2	3fa	Ph	H	4f	85
3	3ka	3-thienyl	$_{\mathrm{H}}$	4k	89
4	3na	n-Bu	Cl	4n	90

^a Conditions: **3** (0.15 mmol), K_2CO_3 (0.77 mmol) in MeOH/ H_2O (4.5/0.1 mL) at rt for 50 min under Ar. ^b Isolated yield (stereoisomer ratio = >20:1).

We studied the scope of nitrogen nucleophiles $\mathbf{2}$ for the reaction of $\mathbf{1a}$; the results are listed in Table 3. A variety of primary amides $\mathbf{2b} - \mathbf{e}$ including N-benzylurea worked well to give the corresponding 3-(amidoalkylidene)oxindoles

Table 3. Pd(0)-Catalyzed Cyclization Reaction of **1a** with **2**^a

entry	2	$ m R^3R^4NH$	3	$yield^b$ (%)
1	2 b	4-CH ₃ C ₆ H ₄ SO ₂ NH ₂ (TsNH ₂)	3ab	94
2	2c	$C_6H_5CONH_2$ (BzNH ₂)	3ac	68
3	2d	$C_6H_5CH_2OCONH_2$ (CbzNH ₂)	3ad	69^c
4	2e	$C_6H_5CH_2NHCONH_2$	3ae	83
5	2f	phthalimide	3af	98^d
6	2g	indolin-2-one	3ag	90
7	2h	oxazolidin-2-one	3ah	64
8	2i	aniline	3ai	44

 a The reaction conditions are the same as those in Table 1 unless otherwise noted. b Isolated yield (stereoisomer ratio = >20:1). c **2d** (0.6 mmol, 3 equiv) and BINAP (4 μ mol, 2 mol %) as the ligand were used. d **2f** (0.6 mmol, 3 equiv) was used.

3ab—**ae** in yields ranging from 68% to 94% (entries 1—4). The cyclization also occurred with secondary amides such as phthalimide (**2f**), indolin-2-one (**2g**), and oxazolidin-2-one (**2h**) to afford the corresponding products in good yield (entries 5—7). The reaction with aniline (**2i**) gave the product **3ai** in 44% yield, together with a side product arising from the direct addition of **2i** to the isocyanato group (entry 8).

In summary, we have developed a palladium-catalyzed amidative cyclization reaction to synthesize 3-(amidoalky-lidene)oxindoles in a highly atom economical manner. The overall reaction accomplishes both intramolecular C—C bond formation and intermolecular C—N bond formation across a carbon—carbon triple bond in a stereoselective fashion. This unique example of cyclization will inspire the development of transition metal-catalyzed reactions producing valuable nitrogen-containing compounds via metallacyclic intermediates.

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Supporting Information Available: Experimental details and spectra data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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